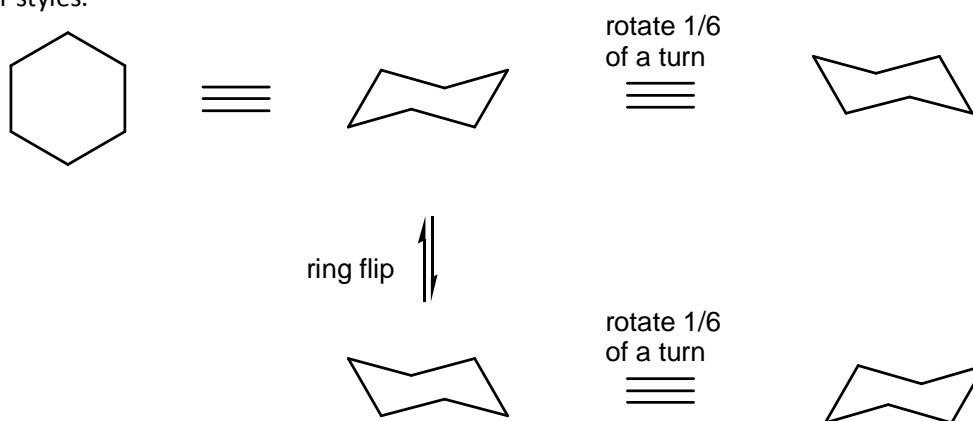


- 1) Using the templates below, draw the flat, top-down view of (1R,3R,5S)-1-bromo-3-ethyl-5-iodo-1-methylcyclohexane. Convert this to one chair conformer, then the other. Show each ring-flip isomer in both chair styles.



- 2) Reacting (S)-4-chlorocyclohex-1-ene with Br_2 and CH_2Cl_2 gives two products. Are these products enantiomers or diastereomers? Based on the two chair conformations of each of these molecules, which product has the most stable conformation possible?

- 3) For the molecule (S)-1,1,2-trimethylcyclohexane, how many gauche interactions exist in each of its two ring-flip forms?