FINAL EXAMINATION

CHEMISTRY 3311

December 17, 1996

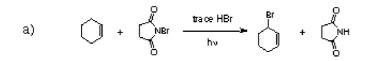
1. (20 points) Provide the missing reagents or products for the following reactions. Carefully show the stereochemistry of each product (if relevant) using wedges and dashes.

a)
$$\longrightarrow$$
 $\xrightarrow{\text{Na.NH}_2(I)}$?

h)
$$\bigwedge^{\circ}$$
 + \bigwedge° $\stackrel{\text{heat}}{\longrightarrow}$?

2. (30 points) Propose a synthesis of the target compounds starting with the substrate provided and any other chemical reagents. Several steps are required in each case. You do not have to show mechanisms for each of the individual steps, but do show the products formed from each of the reactions you perform.

3. (30 points) Write out a detailed, stepwise mechanism for the following transformations. Draw all reasonable resonance structures for any intermediates along the reaction pathway.



b)
$$\frac{Br_2}{cat \ FeBr_8}$$

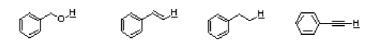
4. (20 points)

A. Draw the Hückel p molecular orbitals for the cyclobutadiene system and arrange them in order of increasing energy. Label the molecular orbitals as bonding, nonbonding, and antibonding as appropriate for their energy levels. Finally, on this diagram show the electron distribution for cyclobutadienyl dication.

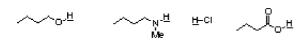
B. Circle the molecules that are aromatic according to Hückel's theory.

5. (25 points) Rank the following [1 - 4 (5), 1 = greatest or most] according to the indicated criteria.

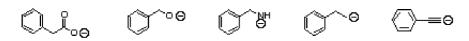
a. Proton acidity



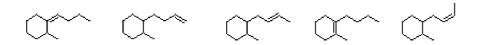
b. Proton acidity



c. Basicity



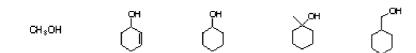
d. Heat of combustion



e. Rate of reaction with tert-butanol

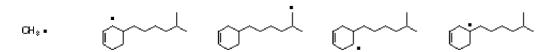
HBr HCl HI HF

f. Rate of reaction with HCl

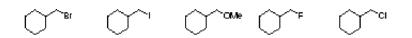


g. Cation stability

h. Radical stability



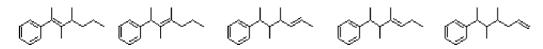
i. Rate of reaction with NaCN



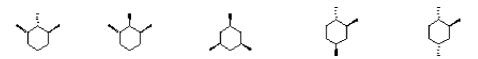
j. Rate of reaction with cyclohexane

F₂ H₂/Pt B-₂ l₂ G₂

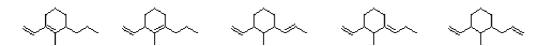
k. Heat of hydrogenation



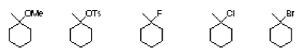
I. Heat of combustion



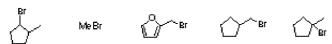
m. Heat of hydrogenation



n. Rate of reaction with potassium tert butoxide



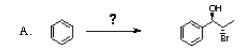
o. Rate of SN2 reaction with sodium acetylide



p. Rate of reaction with cyclohexene

HB⊤ HF HI HCI H₂O

6. (20 points) Propose a synthesis of the target compounds starting with the substrate provided and any other chemical reagents. Several steps are required in each case. You do not have to show mechanisms for each of the individual steps, but do show the products formed from each of the reactions you perform.



7. (25 points) Provide the missing reagents and reaction conditions where necessary for the following reactions.

$$d. \qquad \bigoplus_{F = F}^{Br} \frac{?}{ } \qquad \bigoplus_{F = F}^{Pr} \frac{?}{ }$$

8. (30 points)

a. Label each of the following pairs of structures as homomers, enantiomers, diastereomers, or constitutional isomers, and indicate whether each compound is chiral or achiral using the check boxes.

сно н — он н — он сн _г он	сно н — он но — н сн _г он	Chiral Achiral	Chiral Achiral
Br Br Br Chiral	Br. Br Br Chiral Achiral	HO, OH Chiral Achiral	но Сhiral
Chiral	Br OH Chiral Achiral	H. C=C=CH Me H	Me. Me≠c=c=c H Chiral □ Achiral

b. Draw all of the isomers of C4H6.